
[1]:

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$$S_G \stackrel{\text{def}}{=} |\dot{S}_a + \dot{S}_b + \dot{S}_c| ; \quad (1)$$

-

$$S_A \stackrel{\text{def}}{=} |\dot{S}_a| + |\dot{S}_b| + |\dot{S}_c| ; \quad (2)$$

- *Buchholz's*

$$S_B \stackrel{\text{def}}{=} |\mathbf{I}| |\mathbf{U}| = \sqrt{U_a^2 + U_b^2 + U_c^2} \sqrt{I_a^2 + I_b^2 + I_c^2} , \quad (3)$$

$$\dot{S}_k = \dot{U}_k I_k^* = P_k + jQ_k \quad k = , k \in \{a, b, c\} ;$$

$$|\mathbf{I}| = \sqrt{I_a^2 + I_b^2 + I_c^2} , \quad |\mathbf{U}| = \sqrt{U_a^2 + U_b^2 + U_c^2} \quad (4)$$

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$$\mathbf{U} = \begin{bmatrix} \dot{U}_a \\ \dot{U}_b \\ \dot{U}_c \end{bmatrix} = \begin{bmatrix} U_a e^{j\mathcal{E}_a} \\ U_b e^{j\mathcal{E}_b} \\ U_c e^{j\mathcal{E}_c} \end{bmatrix} , \quad \mathbf{I} = \begin{bmatrix} \dot{I}_a \\ \dot{I}_b \\ \dot{I}_c \end{bmatrix} = \begin{bmatrix} I_a e^{j\mathcal{I}_a} \\ I_b e^{j\mathcal{I}_b} \\ I_c e^{j\mathcal{I}_c} \end{bmatrix} \quad (5)$$

(1) – (3)

$$S_G \leq S_A \leq S_B . \quad (6)$$

(power factor) $\cos \phi_X = P/S_X$ ($X \in \{A, B, G\}$)

$$\cos \phi_G \geq \cos \phi_A \geq \cos \phi_B \quad (7)$$

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$S_A \geq S_B$,

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(1)–(3) (3)

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(1) (3)

$$S_B^2 = S_G^2 + D_u^2 \quad (8)$$

$$S_G^2 = P^2 + Q^2 \quad D_u^2$$

(5)

$$\mathbf{U} \times \mathbf{I} = \begin{bmatrix} \dot{U}_b \dot{I}_c - \dot{U}_c \dot{I}_b \\ \dot{U}_c \dot{I}_a - \dot{U}_a \dot{I}_c \\ \dot{U}_a \dot{I}_b - \dot{U}_b \dot{I}_a \end{bmatrix} \quad (9)$$

$$\mathbf{D} = \mathbf{U} \times \mathbf{I}$$

$$|\mathbf{D}| = |\mathbf{U} \times \mathbf{I}| = D_u \quad (8)$$

$$S_G^2 = P^2 + Q^2,$$

Buchholz's

$$S_B^2 = P^2 + Q^2 + D_u^2,$$

U (5).

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1. // 2009. 6. 22-27: 2009. 7. 15-21, <http://www.kudrinbi.ru> 2. // 1953. 2. 56-61. 3. " " " : (), 2006. promel2000.narod.ru/posobia/tm.doc. 4. // « » VI , EPQ-2008: C . : , 211-214.

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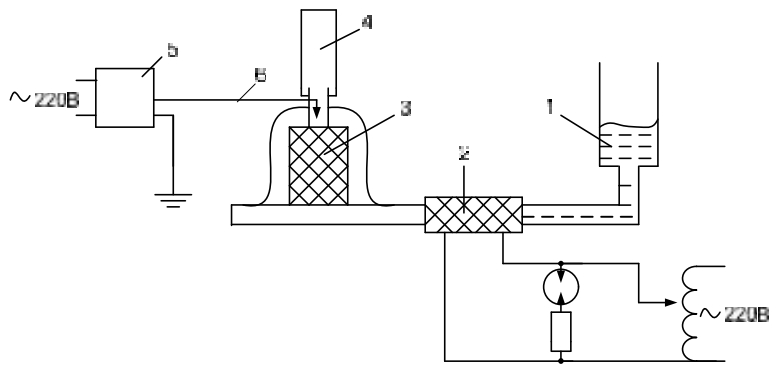
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29 1996 . 488/1513. 6. .2.5-16-99. -

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$$C_p = t \frac{fV_0V}{\ln\left(\frac{a-r}{r}\right)E}, \quad / \quad (2)$$

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$$r = 8,69 \left(\frac{R}{2} \sqrt{\frac{C}{L}} + \frac{G}{2} \sqrt{\frac{L}{C}} \right), \quad / \quad (3)$$

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$$Z = \sqrt{\frac{L}{C}} \quad (4)$$

Z = 100 ± 15
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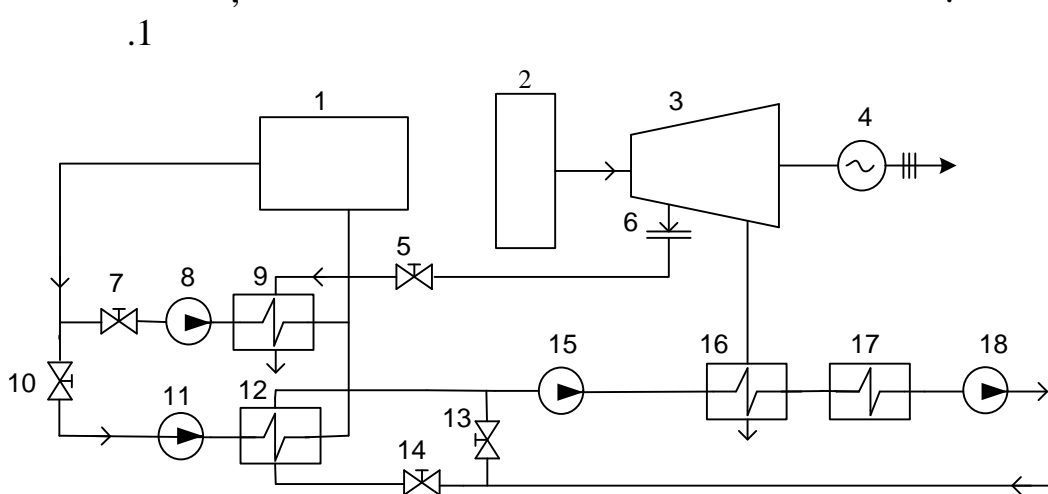
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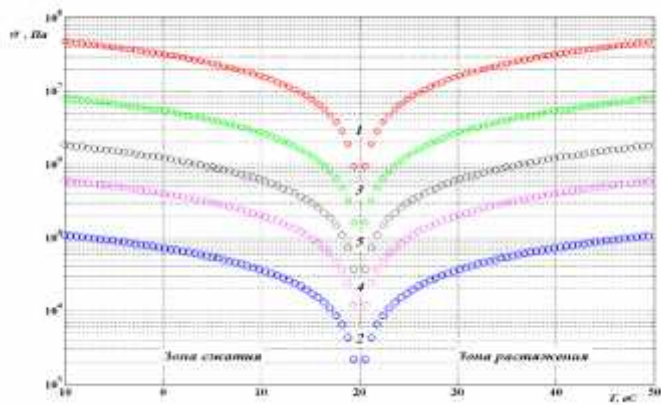
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	1 · 10 ⁻⁴
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	(2.2 – 2.5) · 10 ⁻⁵
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Buchholz's

$$S_B = |U| |I| \quad (1)$$

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$$|I| = \sqrt{I_a^2 + I_b^2 + I_c^2}, \quad |U| = \sqrt{U_a^2 + U_b^2 + U_c^2} \quad (2)$$

(3D-)

$$\mathbf{U} = \begin{bmatrix} \dot{U}_a \\ \dot{U}_b \\ \dot{U}_c \end{bmatrix} = \begin{bmatrix} U_a e^{j\omega t} \\ U_b e^{j\omega t} \\ U_c e^{j\omega t} \end{bmatrix}, \quad \mathbf{I} = \begin{bmatrix} \dot{i}_a \\ \dot{i}_b \\ \dot{i}_c \end{bmatrix} = \begin{bmatrix} I_a e^{j\omega t} \\ I_b e^{j\omega t} \\ I_c e^{j\omega t} \end{bmatrix} \quad (3)$$

(3)

$$\mathbf{U} = \frac{\sqrt{2}}{T} \int_0^T \mathbf{u}(t) e^{-j\omega t} dt, \quad \mathbf{I} = \frac{\sqrt{2}}{T} \int_0^T \mathbf{i}(t) e^{-j\omega t} dt \quad (4)$$

$$\mathbf{u}(t) = \begin{bmatrix} u_a(t) \\ u_b(t) \\ u_c(t) \end{bmatrix}, \quad \mathbf{i}(t) = \begin{bmatrix} i_a(t) \\ i_b(t) \\ i_c(t) \end{bmatrix} \quad (5)$$

3D-

$$\dot{S} = U \cdot I^* = \dot{U}_a I_a^* + \dot{U}_b I_b^* + \dot{U}_c I_c^* \quad (6)$$

$$S_G = |U \cdot I^*| \quad (7)$$

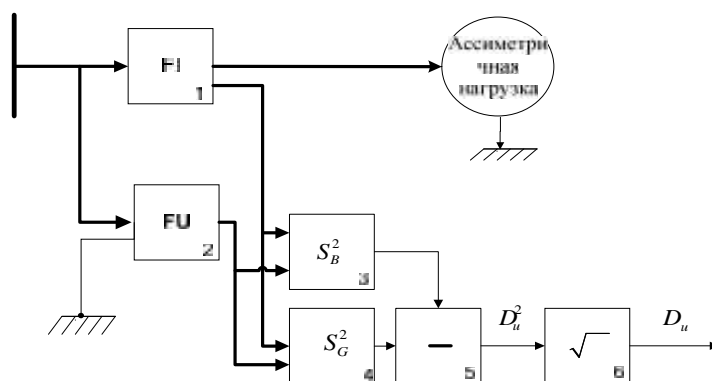
$$S_B \geq S_G$$

$$S_B^2 = S_G^2 + D_u^2 \quad (8)$$

D_u

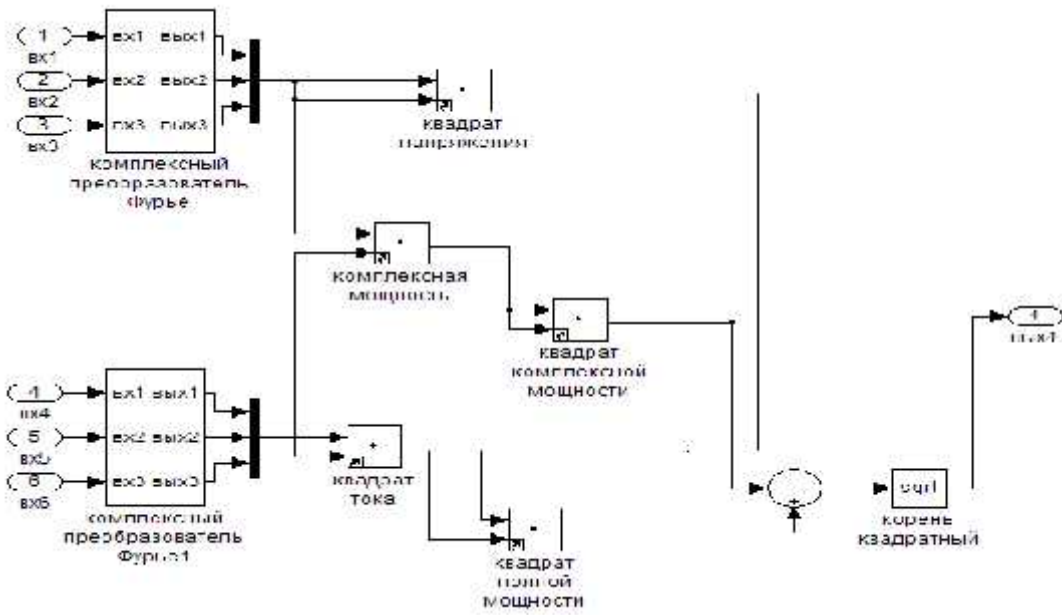
$$D_u = \sqrt{S_B^2 - S_G^2}$$

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Simulink:



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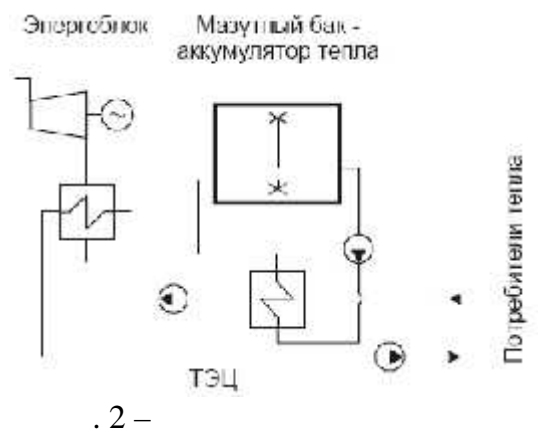
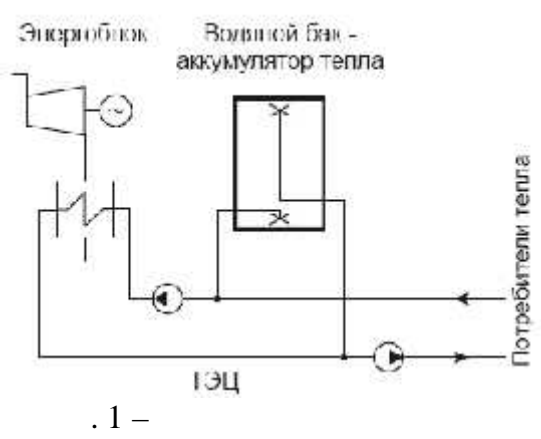
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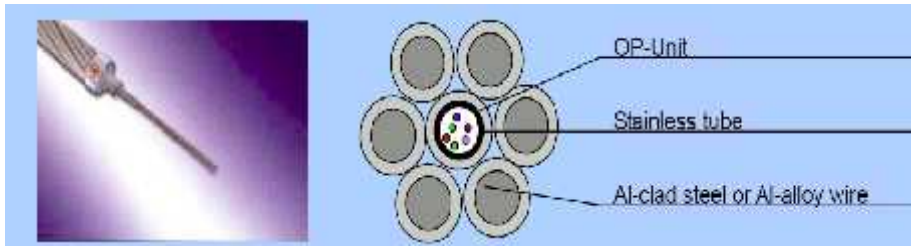
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OPGW

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 - 2. - 25
- 1 ().



1 – OPGW

$$= \pi \cdot R \cdot t \cdot W$$

Q = m \cdot (T - T_0)

$$\frac{300/48}{50} \cdot \frac{240/56}{55} \cdot \left(\frac{I = 25}{90^\circ (150/24)} \cdot \frac{500^\circ (50/8)}{60^\circ} \right)$$

OPGW. 100°

$$\frac{348}{150^\circ} \cdot \left(\frac{I = 25}{300/48 = 25} \right) \cdot 20$$

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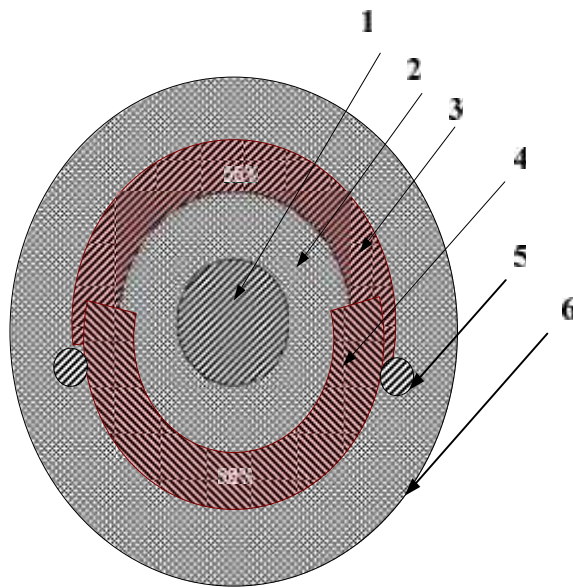
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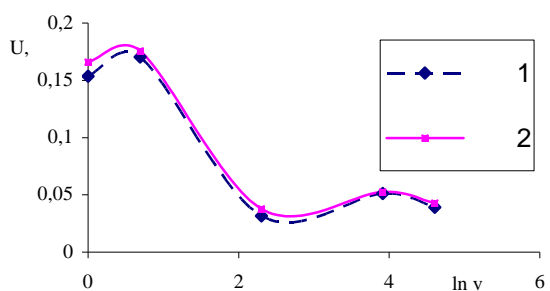
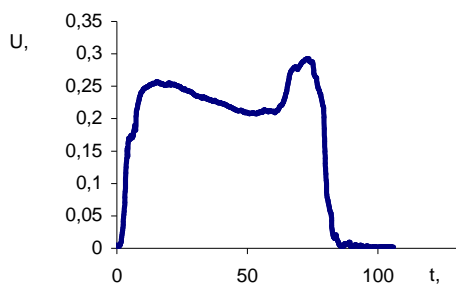
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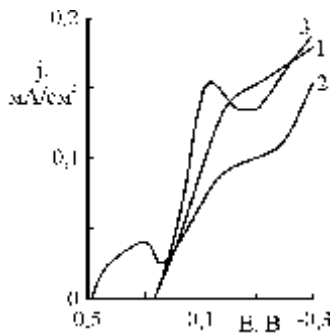
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/ 3.

Ag/Cu = 0,1 ÷ 0,05
 Ag = 0,25 ÷ 0,3
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Cu = -(0,1 ÷ 0,05)

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- lg(1-j/j) (. 2)

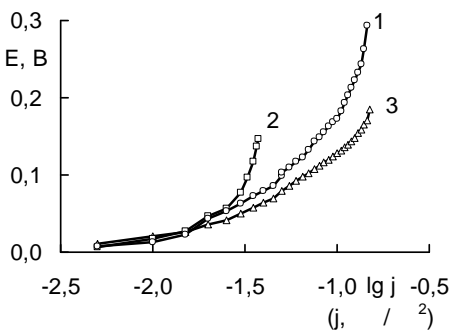
(. 1)
 Cu⁺

Cu²⁺

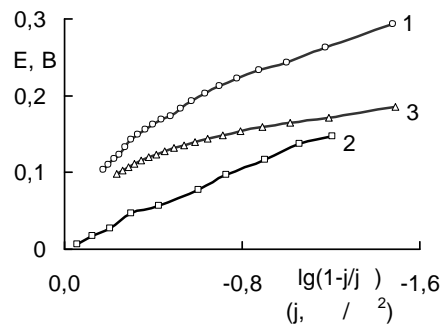
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Ag:Cu=1:10,

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$$\begin{cases} \omega \cdot \frac{dS}{dt} - \frac{K_0}{1 - \frac{c}{c^*}} \cdot S_0 \cdot \gamma^2 = 0 \\ \omega \frac{dC}{dt} + \beta_{ob} \cdot (c^* - c) - \Psi \cdot \frac{K_0}{1 - \frac{c}{c^*}} \cdot S_0 \cdot \gamma^2 = 0 \end{cases}$$

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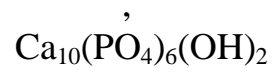
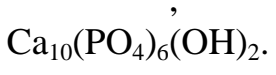
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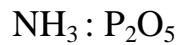
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(, 2 5 ~ 5-8% . , %: CaO-30.55, P₂O₅-13.0, Fe₂O₃-3.50, Al₂O₃-2.79, MgO-1.2, F-1.10, CO₂-3.59, SiO₂-28.1. : , %: HNO₃-4.85, H₃PO₄-8.02, Ca(NO₃)₂-37.89 , Fe(NO₃)₃-3.33, Al(NO₃)₃-3.85, Mg(NO₃)₂-1.98, H₂SiF₆-0.39, NaNO₃-0.08, KNO₃-0.81, 2 -38.75. NP-



$$\frac{dC}{dt} = K \exp\left(\frac{-E}{RT}\right) (1-\Gamma)^n C_k^0 H \quad (2)$$

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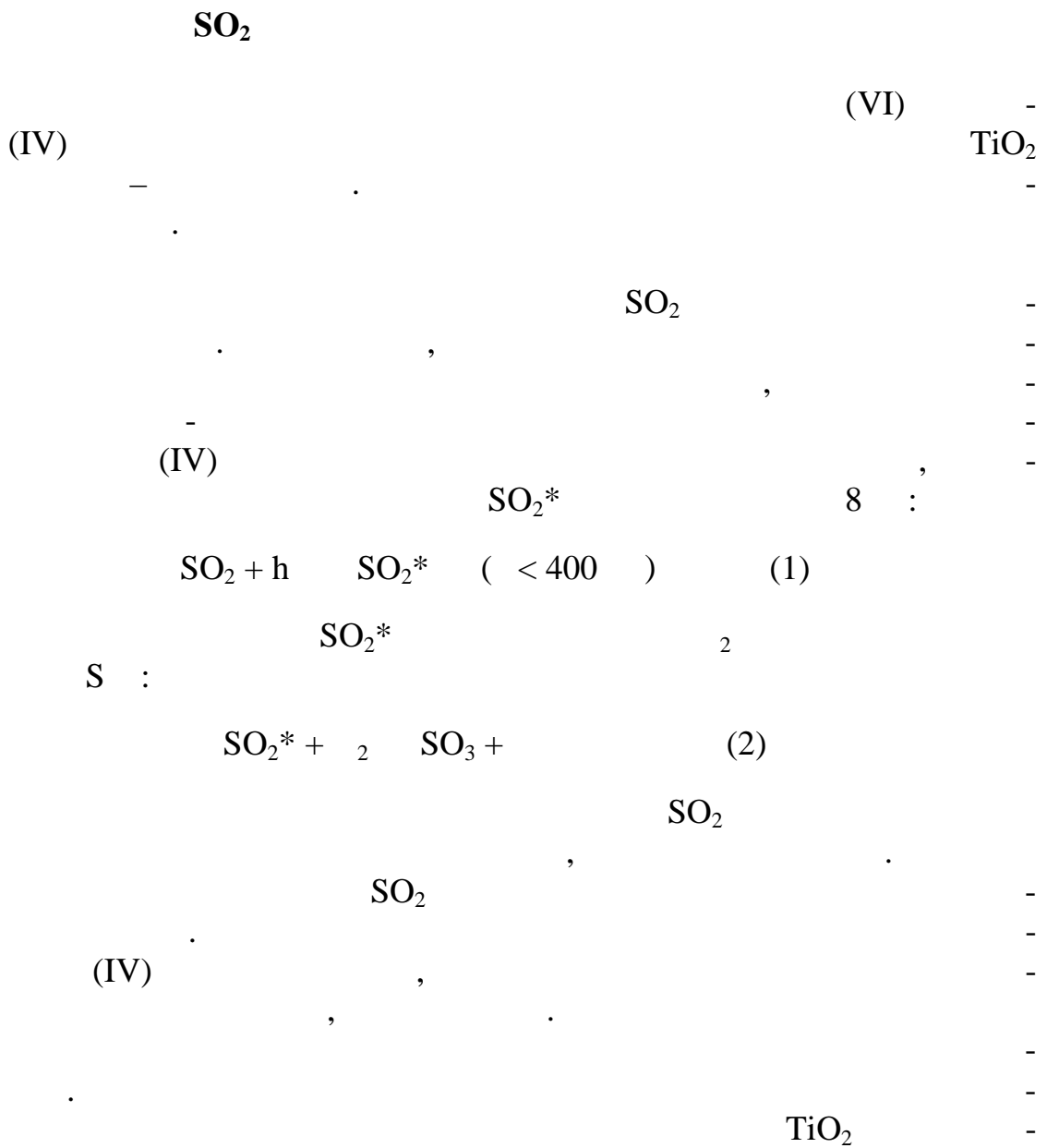
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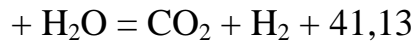
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 -2 [1] -
 : CuO = 40,0±5,0; ZnO = 40,0±5,0; Al₂O₃ -
 , % . -
 = 10,0±2,0. -

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 -2 -
 . _____ :
 , , . -

[2, 3].

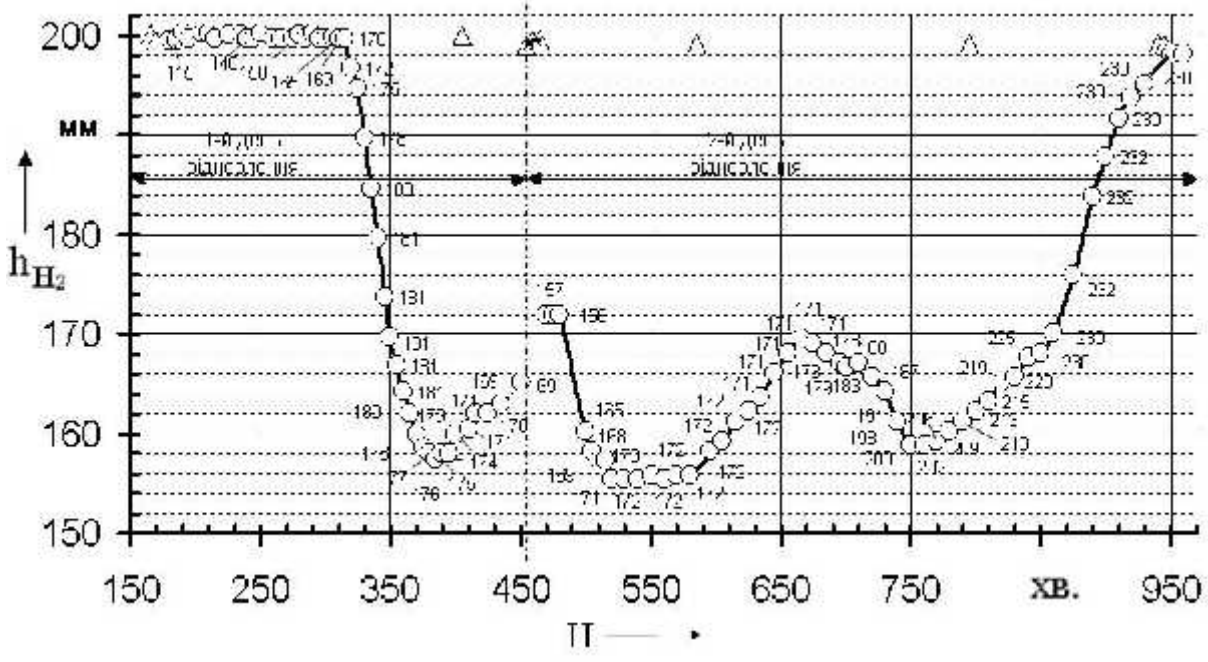
[3].

« » , ()

N₂, .) (, 2, 2,
 () , -

5,5% CuO ()
 = 0,14 CuO, 170-172°
 (.) CuO - Cu -

(,) - 157-158°
 14,6% . ().



$W = 4200^{-1}$.

Δ, \circ - H_2
 \square - H ().

($2 - 2 - N_2$) O_2 [0,14 , -
 (), . . , %: $H_2 = 5,5$; $N_2 -$]

: 1. 6-04687872.047-2000. (- ,
 -2). - 2000.05.12. - : « » - 2000. - 45 . 2.

// , ,
 , 1998.- . 18. - . 136 - 141. 3.
 // . - 1998.- . 39.- 6.- . 869-874.

$$(180 - 350 \text{ m} / \text{)}, \quad ((93-120) \cdot 10^7 \text{ }^{-1}),$$

(PUESTA).

$$20 / \text{ }^2,$$

$$3,5 / \text{ }^2,$$

$$>n \cdot 10^8$$

PUESTA.

06 08 ,

1 3, 1 2

$$i_1 = 1,2 \cdot 10^8$$

$$i_2 = 2,16 \cdot 10^7$$

$$i_1 = 8 \cdot 10^8$$

$$i_3 = 2 \cdot 10^8$$

20°

$$20 - 300 - 20^\circ \quad i_1 = 4,3 \cdot 10^{11}$$

$$i_3 = 2,3 \cdot 10^{10}$$

$$i_1 = 1,14 \cdot 10^{11}$$

$$i_2 = 0,38 \cdot 10^{11}$$

$$t = 620^\circ$$

$$t = 780^\circ$$

1,

$$2 - t = 640^\circ$$

$$t = 820^\circ$$

$$860^\circ$$

1

$$2 \quad t = 880^\circ$$

20°

t =

$$1-1,13 / \text{ }^2, \quad 3-12,17 / \text{ }^2,$$

$$3-17-47-$$

98

$$(20 / \text{ }^2).$$

1 2 30%-

$$0,0056 / \text{ }^2.$$

$$: \quad 1 - 0,5155 / \text{ }^2, \quad 2 -$$

$$(\quad 20-350-$$

20°). - 5. 1 -
3 , 3 - 2 . -

25 25 , Mefrit.

24788. -

4-5 . -

1 2 , -

PUESTA.

1 3 -

621.359.669

_____ . , . , . , . .

Fe - Co

Fe²⁺ , , Fe - Co -

Fe²⁺ Fe³⁺ .

Fe²⁺ -

10⁵ . -

[1], , . -

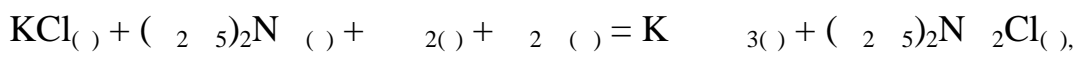
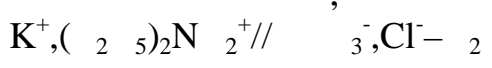
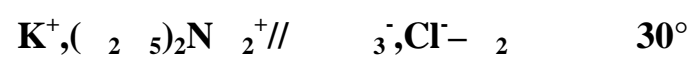
H₃Cit H₂Cit¹⁻ HCit²⁻ Cit³⁻,

$\text{Cit}^{3-} - \text{C}_6\text{H}_5\text{O}_7^{3-} - \text{Cit}^{3-}$ 7 -
 Cit^{3-} , -
 H_2Cit^- , Fe^{2+} 3-4 -
 $[\text{FeHCit}]$ (~2,12). -
 $\text{Fe} - \text{Co}$, -
 10 [2] -
 120 - 150 . .

: 1.
 / : « », 2006. – 272 . 2. Zech N. Anomalous
 Codeposition of Iron Group Metals I. Experimental Results / N. Zech, E.J. Podlaha, D. Landolt // Journal
 of The Electrochemical Society. – 1999. – .146, 8. – .2886–2891.

541.123.61:541.8

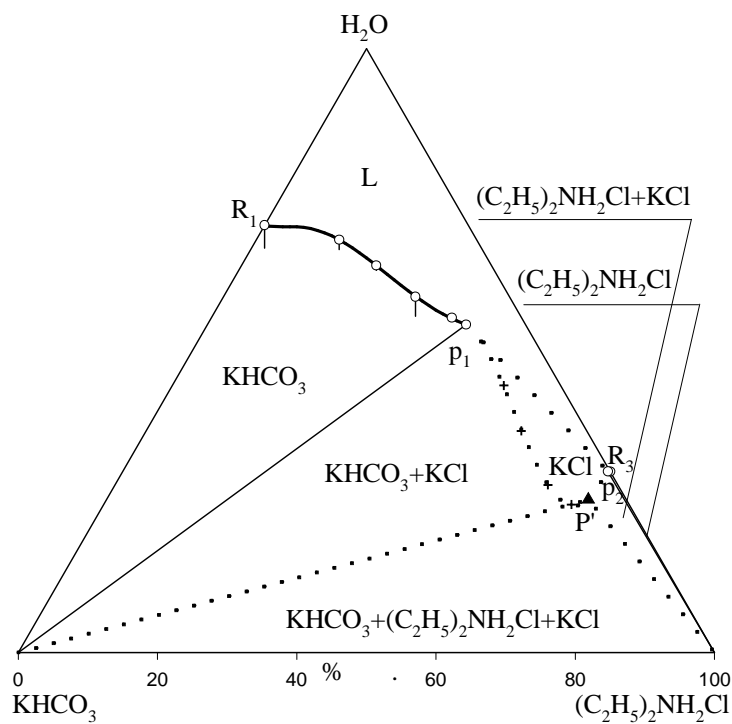
_____



10, 20 25° [1,2].

30±0,1°

[3].



K₂CO₃-(C₂H₅)₂NH₂Cl-H₂O 30°

KHCO₃-KCl-H₂O

(1,),
(%): 8,51 KHCO₃; 37,18 (C₂H₅)₂NH₂Cl; 54,31 H₂O.

(2,)
(%): 0,32 KHCO₃; 69,71 (C₂H₅)₂NH₂Cl; 29,98 H₂O.

KHCO₃-KCl-H₂O
K₂CO₃-(C₂H₅)₂NH₂Cl-H₂O (,) (%): 5,46
KHCO₃; 69,21 (C₂H₅)₂NH₂Cl; 25,33 H₂O.

KHCO₃-KCl-H₂O
K₂CO₃-(C₂H₅)₂NH₂Cl-H₂O , 2

K₂CO₃-(C₂H₅)₂NH₂Cl-H₂O .

K₂CO₃-(C₂H₅)₂NH₂Cl-

2

0,05; 0,1; 0,15

100

0,05

96%;

1,67 / 3 5,2 / 2;

=4,5

_____

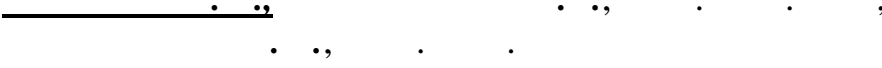
1; 3; 7; 28

28

1; 3; 7; 28

: BaO·Fe₂O₃, 2CaO·Fe₂O₃, 3CaO·2SiO₂·3H₂O, 2CaO·SiO₂·H₂O,
BaO·SiO₂·H₂O, BaO·SiO₂·6H₂O, α - Fe₂O₃, Ca(OH)₂.

621.357.7



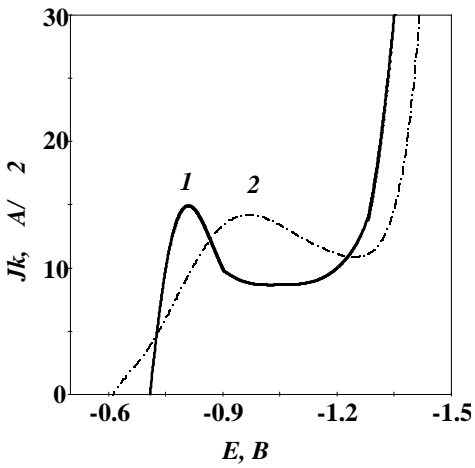
$E_{Sb^{3+}/Sb}^0 = 0,24$) ,

$(E_{Sn^{2+}/Sn}^0 = - 0,14$,

5 / 2 .

0,1 -

. 1,
0,1



. 1 -

Sn (1)

Sn-Sb (2)

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[1-3],
[4].

[5].

[6].

() 28084-94 336

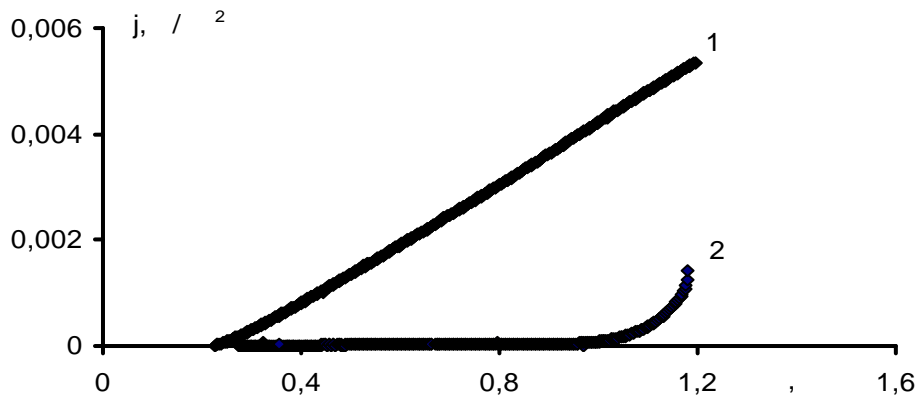
90±2°

(.1, 1)

(.1, 2)

1 ().

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2 -

	Z, %		
	= 0,6	= 0,8	= 1
	99	99	98
	99	98	51
	99	97	93

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.-2004.- . 73, 1. .79. 3.

.-1991- .27, 3.- .379 - 387. 4.

1984.- 400 . 6.

III

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76-78.

Al₂O₃ – Si₃N₄ –

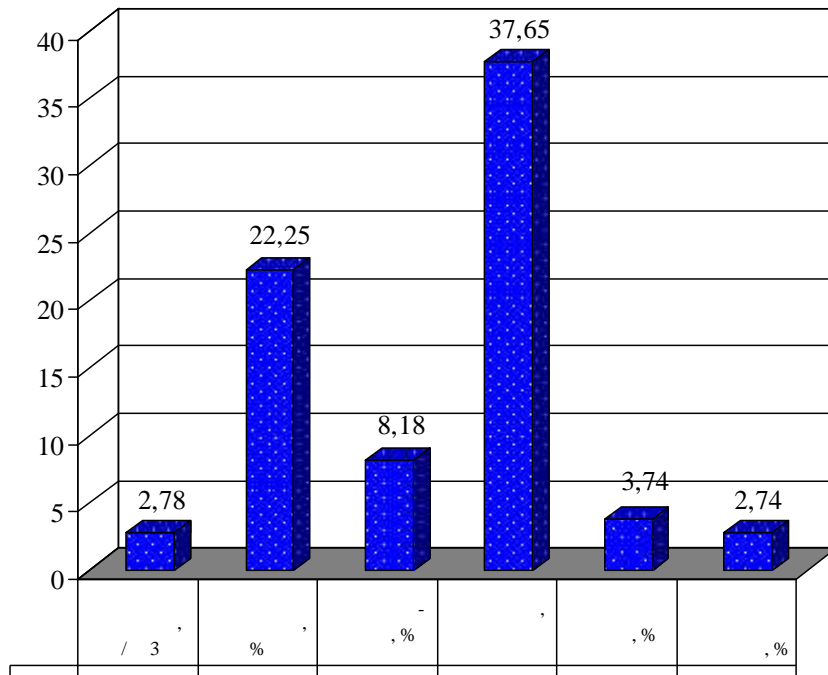
Al₂O₃ – Si₃N₄ – ,

2,6–2,8 / 3,

800 ° – 4–12 %,
– 3,5–10,5 %.

12–38 ,

18–27 %,



5 . % , 0,5 . % 90 . % Al₂O₃, 5 . % Si₃N₄,
(100 %)

621.35

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55 . % (-1) 50 . % (-2).

[1].
[2].

28 3 %- MgSO₄ -
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 28
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 < 0,9 – , -
 (, .) < 0,8 – -
 , (, -
 .) [2]. -
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 7 -1 -2, 3 %-
 MgSO₄, , -
 ()₂, 5 3Si₄O₁₆ (9.
) MgSO₄.
 , , , -
 10. , , , -
 ()₂ , , , -
 ()₂ MgSO₄, -
 SO₄ MgO .
 8 9.

	28 -		-
	3 %- MgSO ₄	2	
-1	26,4	32,6	0,8
-2	61,2	51,0	1,2

, -
 (= 0,8 – 1,2). -
 -1 -
 -2, , MgSO₄ -
 , -
 . , -
 -1.

3 %- MgSO_4 28

(6 15). MgSO_4 6

SO_4 $\text{MgSO}_4 \cdot 6\text{H}_2\text{O}$ 6

26,5 3 %- MgSO_4 -1 73,5
 (- 7), -2

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 7.091606
 8 . 2. / : . . . , . . . : , 1998.-
 : , 1973. - 504 .

669.849

_____ . ” . ” , . ”
 . ” . ”

(W, Mo, Re)

Mo, Re. - , W,

W-Re-

HF HNO₃

(U<0);

(F<0);

-

40 - 50 ° ,

100 - 150

/ ²,

: 85 - 90 %.

(, , ,) ,

100 %.

W-Re-

621.35

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399 °)

9 / .

(245 am)
399 ° .
(316 °),

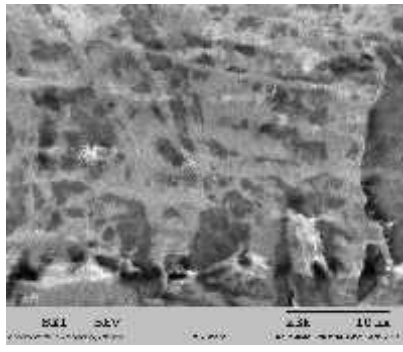
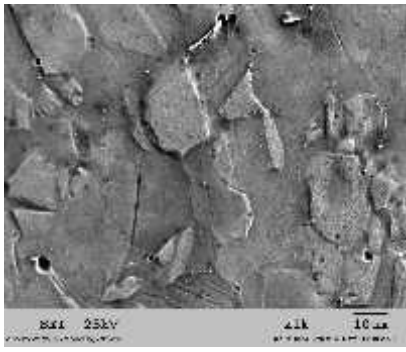
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14 /
 $y=1/2 \cdot \pi \cdot f \cdot Cs$ $x=Rs$

7 /
7,25 22 ° .



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350 ° ,
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R₂O – RO – RO₂ – R₂O₃ – P₂O₅ – SiO₂

– RO – RO₂ – R₂O₃ – P₂O₅ – SiO₂ R₂O
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 ,
 CaO / P₂O₅ 1 3. CaO
 5 8 . %, P₂O₅ – 4 6
 . %.
 .
 Zn₃(PO₄)₂
 ZnO / P₂O₅ = 3.
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 ;
 ,
 .
 2,6,7,8 . % ZnO
 NaCaPO₄. 4 5 . % ZnO
 .
 ZnO 5 – 8 . %
 CaO TiO₂
 ,
 CaO 10 . % 12
 . % ZnO -

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-0,1 - +0,8 .

(-

, -0,1 - +0,1)

(, +0,5 - +0,8

)

(+0,1 - +0,5)

Pd

Pd

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DM 59,

PUESTA.

DM 59
Co, Ni, Cu, Fe, Mn,

(),

DM59-1⁷ DM59-7,

(d)

200 ° ,
d₁=10

2

860 °
d₂ 7

9,0 – 9,7

d

DM 59-1 (d₁=0,9), DM 59-2
(d₂=1,0), DM 59-6 (d₃=0,8), . . .
Cu, Co, Ni.

DM 59-5, CuO DM 59-2, DM 59-4,
Fe, Co, Ni.

12.

664.3 + 543.635.34

(65 %),

547:722:667.633:26

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12±2%.

70⁰ .

c

c

664.3 + 543.635.34

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150 .

50%

1:1.

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503 .

664.3 + 543.635.34

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1:1, 1:2,

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665.3.577

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664.3 + 543.635.34

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663.258.3

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$$\frac{35^\circ}{10'} \frac{45^\circ}{10'} \frac{60^\circ}{15'} \frac{72^\circ}{12'} 30'$$

$$1^\circ / \dots - 1:2. =5,6.$$

$$1=105^\circ \quad 36$$

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: 1, 2, 3%

(4) -

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30

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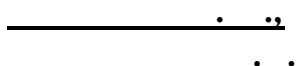
)

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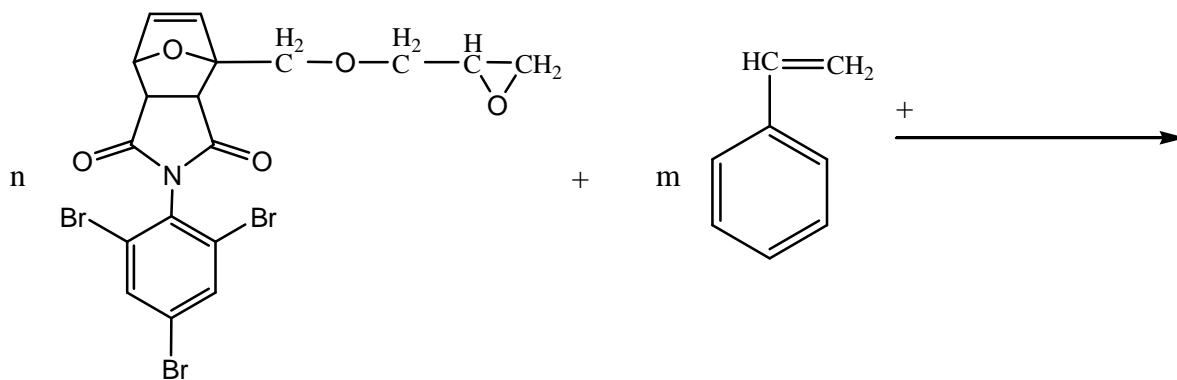
)

(4).

547:722:667.633:26



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120°)
RuCl₃, OsCl₃.
() ()



2

$r_{-}, s_{-}, x_{-}, u_{-}$

664.346

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663.21.3

15 %.

80 %,

3

50-60 %

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678:539.1.074.3

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631.363.25.02

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(1):

$$W \nabla T = \chi \cdot \Delta T;$$

$$(W \nabla) W = -\nabla \frac{P}{\rho} + \nu \Delta W;$$

$$\operatorname{div} W = 0,$$

(1)

, V -
, W -

(1)

1)

2)

(1):

$$P_{\text{сум}} = 0,148 \cdot \rho \cdot W_c^2 \cdot \frac{H}{t} \left\{ \frac{[\operatorname{Re}_c - XA + 36,9 \cdot (X \cdot A)^{0,625}]^{0,3}}{\operatorname{Re}_c \cdot \cos \varphi \cdot \cos \alpha} + 3 \cdot \varphi^2 \cdot \sin \varphi \right\}$$

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- : « », 1977 . 2.
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678:614.841.332

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80%(250 .³)

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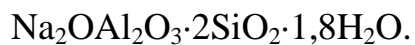
45 55 / .
40 50 / .
10-70 / . 50-65 /
9-11 / 75%.
11,5-12%, - 1,33-1,36%. 0,5 .

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« 2009 . 2. » (14-16 2009), 3, . 144-145, ,
45860.
. 25.11.2009 . 13D 1/00. . . .

678:549.67

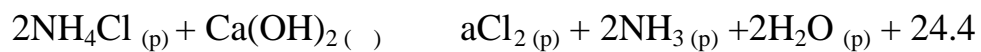
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4-40

661



30 %

Ca(OH)₂,

NH₄ 1

Ca(OH)₂.

30 %

50-60 %

14,88 %

6,03 . .

33,5 / ³

18,43 / ³,

12,7 % 7 %.

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65.012.8

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Fragmented text consisting of various punctuation marks (commas, dashes, colons, semicolons) and some faint characters scattered across the page.

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: " , 2007, N 10). 3. -

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65.014.1:331.101.3

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– 2- , 2002 – 888 . 4.

: http://www.auditfin.com/fin/2000/2/upr_fin/uprfin1.asp#_Toc478045386 5.

/ 2- – , 2004.

[2, . 169].

: **1.** . . . / . . . ,
 - . . . : , 2006. - 544 . **2.** . . . , . . . / -
 // . - 2008. - 4. - . 163-173. **3.** . . . -
 2009. - 7. - . 89-93. **4.** . . . : . . .
 . - . . . : , 2001. - 704 . **5.** -
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[1].

: 1) ' - ; 2) ' - ; 3)

ABC.

„Direct costing” -

„Standard costing” -

ABC (Activity Based Costing)

- Absorption costing -

- Target costing -

- Kaizen costing -

Target costing.

[4].

[2].

[3].

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